

**41.5** A cooling tower cools 150gpm of condenser water maintaining a 15°F delta T. On a design day, air enters the cooling tower at 90°F dry bulb and 75°F wet bulb and leaves at 105°F dry bulb and 80% relative humidity. What volume flow rate of air through the cooling towers is required?

- A. 8100cfm
- B. 12,500cfm
- C. 69,400cfm
- D. 119,700cfm

The quantity of heat removed from the air is added to the condenser water. Therefore:

$$Q_a = Q_w$$

The sensible heating of the condenser water can be determined using the rule of thumb, where the flow rate and differential temperature are known:

$$Q_w = 500gpm\Delta T = 500(150)(15) = 1,125,000 \frac{Btu}{hr}$$

The heat removed from air in a cooling tower is driven by a substantial amount of evaporation; therefore, use the rule of thumb for *total* cooling of air (including both sensible and latent) should be used:

$$Q_a = 4.5cfm\Delta h$$

where the volume flow rate of air is unknown. To specify the change in enthalpy, use the **psychrometric chart** to look up the enthalpy values for entering and leaving conditions, both of which are fully defined. Use of the **high temperature** psychrometric chart is required for State 2.

State 1: Entering Air	State 2: Leaving Air
$T_{1,db} = 90^\circ F$	$T_{2,db} = 105^\circ F$
$T_{1,wb} = 75^\circ F$	$T_{2,wb}$
$\phi_1$	$\phi_2 = 80\%$
$h_1 = 38.4 \frac{Btu}{lb}$	$h_2 = 69.4 \frac{Btu}{lb}$

Calculate the change in enthalpy:

$$\Delta h = h_2 - h_1 = 69.4 \frac{Btu}{lb} - 38.4 \frac{Btu}{lb} = 31 \frac{Btu}{lb}$$

Use the fact that  $Q_a = Q_w$  to solve for the unknown volume flow rate. Note that by using the rule of thumb selected, the units will automatically come out in *cfm* provided the enthalpy is in  $\frac{Btu}{lb}$  and the heat transfer is in  $\frac{Btu}{hr}$ .

$$cfm = \frac{1,125,000}{(4.5)(31)} = 8065cfm$$

**Answer A**

**41.6 Air enters an evaporative cooler at  $85^\circ F$  and 30% relative humidity and exits at  $75^\circ F$ . What is the saturation efficiency?**

- A. 37%
- B. 47%
- C. 53%
- D. 63%

The **saturation efficiency** of an evaporative cooler is defined by the formula:

$$\varepsilon = 100 \frac{(T_1 - T_2)}{(T_1 - T'_s)}$$

where  $T_1$  is the entering air dry bulb temperature,  $T_2$  is the leaving air dry bulb temperature, and  $T'_s$  is the wet bulb temperature of the entering air. The entering wet bulb temperature is the minimum possible leaving air temperature achievable, such that for a 100% efficient process,  $T_2 = T'_s$ .

Since both the entering and leaving dry bulb temperatures are known, the only additional information to be determined is the entering wet bulb, which can be pulled from the **psychrometric chart** since the entering state is fully defined:

$$T_{1,db} = 85^\circ F$$

$$\phi_1 = 30\%$$

$$T_{1,wb} = 63.6^\circ F$$

Calculate the saturation efficiency:

$$\varepsilon = 100 \frac{(T_1 - T_2)}{(T_1 - T'_s)} = 100 \frac{(85^\circ F - 75^\circ F)}{(85^\circ F - 63.6^\circ F)} = 47\%$$

**Answer B**