

36.28 In a refrigeration cycle, R-134a leaves the evaporator at 20psia with 20°F of superheat. What is the entropy of the refrigerant entering the compressor?

- A. $0.03 \frac{\text{Btu}}{\text{lb}^\circ\text{F}}$
- B. $0.22 \frac{\text{Btu}}{\text{lb}^\circ\text{F}}$
- C. $0.24 \frac{\text{Btu}}{\text{lb}^\circ\text{F}}$
- D. $106 \frac{\text{Btu}}{\text{lb}^\circ\text{F}}$

In the Reference Handbook, search for **Properties of Refrigerants** and use the phrase: **Pressure Versus Enthalpy**. Find the chart for Refrigerant 134a.

Following a horizontal line at a pressure of 20psia to the rightmost edge of the saturation curve, corresponding to a saturated vapor condition. Continue horizontally to the right to account for the additional 20°F of superheat. Note that to the right of the saturation curve, lines of constant temperature are curved downward (nearly vertical). This is the compressor entering condition, typically noted as State 1 of the refrigeration cycle.

Once the state point has been identified, read off the entropy from the chart. Note that lines of constant entropy run roughly southwest/northeast on the diagram.

$$s \approx 0.235 \frac{\text{Btu}}{\text{lb}^\circ\text{F}}$$

Answer C